M1. (a) (i) alpha (particle)

(ii) (unstable) nucleus
 accept (unstable) nuclei
 do not accept middle
 do not accept helium nucleus

(iii) same number of protons

 accept same number of electrons
 accept same atomic / proton number
 accept they both have <u>92</u> protons
 same number of neutrons negates answer

(b) (i) 4500 million years do **not** accept 4500 years

(ii) curve starting at 100 000 with a correct general shape

passing through (4500, 50 000) and (9000, 25 000) allow **1** mark for points plotted **or** line passing through (4500, 50 000) and (9000, 25 000)

[6]

1

1

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1

M2.	(a)	(i)	number of protons are the same accept atomic number / number of electrons for number of protons	1
			number of neutrons are different accept mass numbers are different – only if the first mark is awarded	1
		(ii)	an electron from the nucleus <i>both parts needed</i>	1
	(b)	deca	ays at the same rate as it is made accept decays as fast as it is made accept absorbed / used by plants (in CO₂) at same rate as it is being made	1
	(c)	(i)	3500 no tolerance	1
		(ii)	adjusted age correctly obtained from the graph accept values between 3700–3800 inclusive accept their (c)(i) used correctly to obtain an adjusted age from the graph	1
			adjusted age +50 second mark can only be scored if first mark awarded if no working shown an answer between 3750–3850 inclusive scores both marks note: any line or mark made on the graph counts as working out	1

МЗ.		(a)	 (i) (total) number of protons plus neutrons accept number of nucleons accept amount for number do not accept number of particles in the nucleus 	1
		(ii)	number of neutrons decreases by one	1
			number of protons increases by one accept for both marks a neutron changes into a proton	1
	(b)	(i)	²⁰⁸ Th 81	1
			correct order only	1
		(ii)	the number of protons determines the element accept atomic number for number of protons	1
			alpha and beta decay produce different changes to the number of proto there must be a comparison between alpha and beta which is more than a description of alpha and beta decay alone or alpha and beta decay produce different atomic numbers ignore correct reference to mass number	ons
				1

[7]

[7]

M4. (a) beta

alpha: would not pass through (the aluminium / foil)

gamma: no change in count rate when thickness changes must be a connection between detection / count rate / passing through and change in thickness

(b) foil thickness increases then decreases (then back to normal / correct thickness) a description of count rate changes is insufficient

1

1

2

1

11

1

gap between rollers decreases, then increases (then back to correct size)**or**pressure from rollers increases then decreases

accept tightness for pressure answers may link change in thickness and gap width for full credit ie: foil thickness increases so gap between rollers decreases (1) foil thickness decreases so gap between rollers increases (1)

(c) 56 (years)

accept any value between 55-57 inclusive allow **1** mark for correct calculation of mass remaining as 1.5 (micrograms) allow **1** mark for a mass of 4.5 micrograms plus correct use of graph with an answer of 12 maximum of **1** compensation mark can be awarded

[7]

М5.	(a)	78
	(b)	atomic I
	(c)	(i) 131 correct order only
		54
		 (ii) 32 (days) allow 1 mark for showing 4 half-lives provided no subsequent step
		 (iii) limits amount of iodine-131 / radioactive iodine that can be absorbed accept increases level of non-radioactive iodine in thyroid do not accept cancels out iodine-131
		so reducing risk of cancer (of the thyroid) accept stops risk of cancer (of the thyroid) 1

[8]

gamma (radiation) will pass through food / packaging this can score if technetium chosen

long half-life so level of radiation (fairly) constant for (a number) of years this can score if strontium / caesium is chosen accept long half-life so source does not need frequent replacement accept answers in terms of why alpha and beta cannot be used gamma kills bacteria is insufficient 1

1

1

1

1

2

 (b) (i) people may link the use of radiation with illness / cancer accept (they think) food becomes radioactive accept (they think) it is harmful to them 'it' refers to irradiated food

(ii) not biased / influenced (by government views)

(iii) any two from:

- data refers only to (cooked) chicken
- data may not generalise to other foods
- the content of some vitamins increases when food / chicken is irradiated
- no vitamins are (completely) destroyed
- (only) two vitamins decrease (but not significantly) accept irradiated chicken / food contains a higher level of vitamins marks are for the explanation only

- (iv) so can choose to eat / not eat that (particular) food accept irradiated food may cause health problems (for some people) accept people may have ethical issues(over eating irradiated food)
- (c) (i) electron from nucleus / neutron **both** parts required
 - (ii) 90 years allow **1** mark for showing 3 half-lives

[11]

1

1

2

- **M7.** (a) (i) any **one** from:
 - nuclear power (stations)
 accept nuclear waste
 accept coal power stations
 - nuclear weapons (testing)
 accept nuclear bombs / fallout
 - nuclear accidents

 accept named accident, eg Chernobyl or Fukushima
 accept named medical procedure which involves a
 radioactive source
 accept radiotherapy
 accept X-rays
 accept specific industrial examples that involve a radioactive
 source
 nuclear activity / radiation is insufficient
 smoke detectors is insufficient
 - (ii) (radioactive decay) is a random process accept an answer in terms of background / radiation varies (from one point in time to another)

(b) any **one** from:

- (maybe) other factors involved accept a named 'sensible' factor, eg smoking
- evidence may not be valid
 accept not enough data
- may not have (a complete) understanding of the process (involved)

1

1

1

1

1

(c) (i) 2

(ii) 218 correct order only

84		
		1

(d) 3.8 (days)

allow **1** mark for showing correct method using the graph provided no subsequent steps correct answers obtained using numbers other than 800 and 400 gain **2** marks provided the method is shown

2

1